

3 F. Proteins a. <30% of Total Calorie Intake
 b. Essential Fatty Acids Clinical Note, p.835

1. Sources in the Diet
 - a. Nine Essential Amino Acids = Histidine, Isoleucine, Leucine, Lysine, Methionine, Phenylalanine, Threonine, Tryptophan, and Valine
 - b. Remaining Eleven Naturally Occurring Amino Acids are Non-Essential
 - c. Complete Protein Food Contains Adequate Amounts of All Nine Essential Amino Acids = "Meat Group" - Meat, Fish, Poultry, Milk, Cheese, and Eggs
2. Uses in the Body
 - a. Collagen and Structural Proteins
 - b. Contractile Apparatus of Muscle Cells
 - c. Enzymes and Functional Proteins
 - d. Plasma Buffers
 - e. Hemoglobin = O₂ Transport
 - f. Membrane Proteins
 - g. Immune Response Compounds
3. Recommended Amounts
 - a. 0.8 g / Kg Body Weight - About 12% of Total Kcals.
 - b. Balance of Several Incomplete Foods can Provide all Essential Amino Acids
 - c. Nitrogen Balance - When Amount of Nitrogen from Protein Ingested = Amount of Nitrogen Excreted from Body

4 G. Vitamins Table 25.2, p.837-838

1. Required for Normal Metabolism in Small Amounts
2. Many Function as Coenzymes
3. Many Heat Sensitive Clinical Note, p.838
4. Fat Soluble = A,D,E, and K
 - a. Can be Stored
 - b. Hypervitaminosis Possible
5. Water Soluble = B Complex and C, Cannot be Stored Predict Quest. 1

5 H. Minerals Table 25.3, p.839

1. Inorganic Nutrients
2. Necessary for Normal Metabolism
3. 4-5 % of Total body Weight
4. Adequately Provided in a Balanced Diet Clinical Note, p.839

II. Metabolism, p. 840 Fig. 25.3, p.841 TA-520

A. Anabolism - Energy-using Processes by Which Small Molecules Made into Larger Molecules

	B. Catabolism - Energy-Releasing Processes by Which Large Molecules Broken into Smaller Molecules		
	C. Cellular Respiration = Cellular Metabolism		
	D. ATP = Energy "Currency" of the Cell	Fig. 25.2, p.840	TA-519
	III. Carbohydrate Metabolism, p. 840		
6	A. Glycolysis		
	1. Input of ATP for Phosphorylation	Fig. 25.4, p.842-843	TA-521, 522
	2. Sugar Cleavage		
	3. Reduced Nicotinamide Adenine Dinucleotide (NADH) Production		
	4. ATP and Pyruvic Acid Production	Table 25.4, p.844	TA-533
9	B. Anaerobic Respiration	Fig. 25.5, p.844	TA-523
	1. Absence of O ₂		
	2. Lactic Acid Production		
	3. Oxygen Debt Created		
	4. Net Gain of 2 ATP		
	C. Aerobic Respiration		
	1. Acetyl Coenzyme A (Acetyl-CoA) Formation with NADH Production	Fig. 25.6, p.845	TA-524
7	2. Citric Acid Cycle		
	a. Direct ATP Production		
	b. Production of NADH and Reduced Flavin Adenine Dinucleotide (FADH ₂)		
	c. Carbon Dioxide Production		
8	3. Electron Transport Chain	Fig. 25.7, p.846	TA-525
	a. Oxidation of NADH and FADH ₂		
	b. O ₂ as Final Electron Acceptor		
	c. H ₂ O Formed		
	d. H ⁺ Movement Leads to ATP Formation ñ Chemiosmotic Model	Predict Quest. 2	
9	4. Summary of ATP Production from One Molecule of Glucose	Clinical Note, p.847	
	a. Aerobic Respiration Yields 36 or 38 Molecules of ATP (Depending on Cell Type)		
	b. 6 CO ₂ and 6 H ₂ O Also Produced		
10	IV. Lipid Metabolism, p. 847	Fig. 25.8, p.848	TA-526
	A. Long-Term Storage for Energy		
	1. 99% Lipid, Primarily Triacylglycerols		
	2. 1% Glycogen (Shorter-Term)		
	B. Free Fatty Acids in Blood Usage Form		
	C. Beta-Oxidation of Fatty Acids to Acetyl-CoA		
	D. Ketogenesis	Clinical Note, p.847	
	1. Formation of Ketone Bodies when Excess Acetyl-CoA Present		
	2. Converted Back to Acetyl-coA in Skeletal Muscle and Used in Citric Acid Cycle		
11	V. Protein Metabolism, p. 847		
	A. Amino Acids (AA) used to synthesize Needed Proteins		
	B. Biosynthesis of Non-Essential AA		
	1. From Keto Acids	Fig. 25.10, p.849	TA-528
	2. Transamination	Fig. 25.11, p.850	TA-529
	C. Amino Acids as an Energy Source	Fig. 25.9, p.849	TA-527

c. Muscles of Respiration
2. Aspect of Energy Usage Over Which Most
Conscious Control

- 15 IX. Body Temperature Regulation, p. 853
- A. Humans as Homeotherms
 - 1. Enzymes are Temperature Sensitive
 - 2. Body Temperature Kept within Narrow Range
 - B. Result of Balance Between Heat Production and Heat Loss Fig. 25.16, p.856 TA-534
Predict Quest. 3
Predict Quest. 4
 - C. Mechanism of Heat Loss Fig. 25.15, p.850
 - 1. Radiation
 - 2. Conduction
 - 3. Convection
 - 4. Evaporation
 - D. Thermoregulatory Control Center in Posterior Hypothalamus Clinical Focus,
Maintains Core Body Temperature Set-Point p.857

IMPORTANT CONSIDERATIONS: If there is only one lecture session available for these topics, stress the highlights of the energy relationships and the relationship between metabolism and body temperature. If three sessions are available, use one for review of nutrients and their routes of entry into the body, one on cellular metabolism and one on metabolic states, metabolic rates and body temperature regulation.

The details of the catabolism of organic molecules are found in Chapter 24. Have students review this material if necessary. Discuss the meaning of "essential" nutrients as a foreshadowing of the interconversion processes to be discussed later in this chapter.

Students may not be familiar with heat as a form of metabolic waste, so the relationship between metabolic rate and body temperature may not be immediately apparent to them. Humans are homeothermic endotherms. Students should be asked to relate the maintenance of a constant body temperature (usually higher than the environment) to homeostasis and efficiency of metabolism.

SEE INSTRUCTOR'S MANUAL AND COURSE SOLUTIONS MANUAL FOR ADDITIONAL RESOURCES.